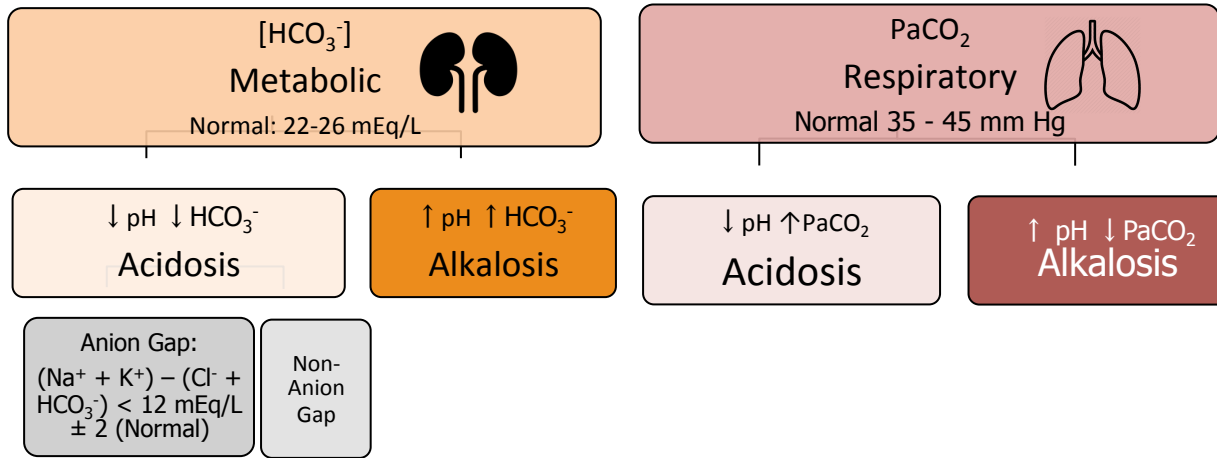


**ABG Interpretation**

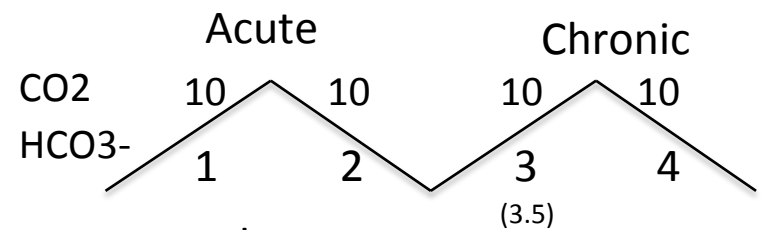


Important Definitions  
 -"emia" change in H+ [] (simply the change in pH)  
 -"osis" the process changing H+ [] (you have Diagnosis)

- ← **1- Does the patient have an acidosis or an alkalosis?**  
 Look at the pH
- ← **2- What is the primary problem – metabolic or respiratory?**  
 Look at the pCO<sub>2</sub>  
 Think ROME = Respiratory Opposite, Metabolic Equal  
 (If pCO<sub>2</sub> change is in the opposite direction of the pH change, the primary problem is respiratory)

- ← **3- Is there any compensation by the patient? Do the calculations.**
- For a primary respiratory problem, is the pH change completely accounted for by the change in pCO<sub>2</sub>  
 if yes, then there is no metabolic compensation  
 if not, then there is either partial compensation or concomitant metabolic problem
- For a metabolic problem, calculate the expected pCO<sub>2</sub>  
 if equal to calculated, then there is appropriate respiratory compensation  
 if higher than calculated, there is concomitant respiratory acidosis  
 if lower than calculated, there is concomitant respiratory alkalosis
- \*\*If it is a metabolic acidosis, you need to do more work = ANION GAP

**Resp Tree Compensation Short Cut**



Example:

pH / CO<sub>2</sub> / O<sub>2</sub> / HCO<sub>3</sub><sup>-</sup>  
 7.34 / 51 (40) / 68 / 27(24)  
 N 11 up 3 up  
 = chronic resp acidosis

	Metabolic Acidosis	Metabolic Alkalosis	Respiratory Acidosis	Respiratory Alkalosis
Compensation	↓ PaCO <sub>2</sub>	↑ PaCO <sub>2</sub>	↑ HCO <sub>3</sub>	↓ HCO <sub>3</sub>
	1 PCO <sub>2</sub> down for every 1 HCO <sub>3</sub> down	PCO <sub>2</sub> up 0.5-0.7 for every 1 HCO <sub>3</sub> up	<u>Acute</u> HCO <sub>3</sub> <sup>-</sup> up by 1 mEq/L for every 10 PCO <sub>2</sub> up <u>Chronic</u> HCO <sub>3</sub> <sup>-</sup> up by 3.5 mEq/L for every 10 PCO <sub>2</sub> up	<u>Acute</u> HCO <sub>3</sub> <sup>-</sup> down by 2 mEq/L for every 10 PCO <sub>2</sub> down <u>Chronic</u> HCO <sub>3</sub> <sup>-</sup> down by 4 mEq/L for every 10 PCO <sub>2</sub> down
Example	<u>Anion Gap:</u> M ethanol U remia D iabetic ketoacidosis P araldehyde I ron, isoniazid L actate E thylene glycol S alicylates  <u>Non-Anion Gap:</u> Hyperchloremic metabolic acidosis  GI loss of HCO <sub>3</sub> • Diarrhea, NEC, small bowel drainage/fistula Renal loss of HCO <sub>3</sub> • renal tubular acidosis (RTA), interstitial nephritis, early renal failure, CA inhibitors • Administration of HCl or other chloride-containing substances (i.e. NS) hyperalimantation, acetazolamide	<u>Chloride-responsive</u> (responds to NaCl/ KCl)  (Diuretics, corticosteroids, gastric suctioning, vomiting)  <u>Chloride-resistant</u> (Any hyperaldosterone state (e.g., Cushing's syndrome), severe K+ depletion )	Breathing Slow  Respiratory Acidosis = Respiratory failure  • Central nervous system depression • Muscle dysfunction • Disease of lungs and/ or upper airway	Breathing Fast  Voluntary hyperventilation • Hypoxemia (includes altitude) • Liver failure • Anxiety • Hyperventilation syndrome • Any acute pulmonary problem  • Acute pulmonary embolism, pneumonia, mild asthma attack, mild pulmonary edema